Physics 100	0		
Electricity	Notes,	Part	I

I.	General	Princip	les
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I.	General Principles
	A. Matter is made of atoms
	B. Atoms are made of protons, neutrons and electrons
	C. These particles that make up the atom have an electric charge.
	Proton +
	Neutron O (neutral)
	Electron
	D. Electric charges exert a force on one another.
	Two positive charges will repel each other.
	Two negative charges will repeleach other.
	A positive charge and a negative charge will Attract each other.
	E. If an object is "charged" we know that it has either an of
	electrons or a lack of electrons.
	F. A neutral object may have positive and negative charges, but itsnet charge is zer
II.	Units
	A. Force Newton N
	B. Charge Coulomb C
	C. Charge on 1 electron -1.6 × 10 -19 C
	D. Charge on 1 proton 1.6 × 10 - 9 C
	E. 1 Coulomb (6.25 × 10 18 clectrons) the anount
	of charge passing through a 100 unit bulb
	in about 1 Second:
	-0.000000000000000000000000000000000000

9x10 Nm2 changes  (III.) Coulomb's Law y 992
III.) Coulomb's Law
9×10° Nm² changes  (III.) Coulomb's Law y 2.92  A. $F = k$ 2.92  between changes (1)²  between changes (2) (1)²
B. Coulomb's Law is an inverse square law
If the distance between two charges doubles, the force decreases by
If the distance between two charges triples, the force distance by 1/9.
IV. Conservation of Charge
A. Electrons are neither <u>created</u> nor <u>destroyed</u>
B. Within a system, the net charge does not change.
Example: A closed system comprises object A (charge = 2C) and object B (charge =-3C)
1. What is the net charge in the system? $2C + (-3c) = (-16)$
2. Which object has more electrons? 3
3. If object A's charge changes to 4C, what is object B's new charge? $4C + x = -1C$ $x = -5C$
V. Conductors and Insulators
A. In some kinds of matter, <u>electrons</u> are not anchored to their atoms'
nuclei; they can "roam around" to some extent. These materials are called
conductors. Metals are typically good conductors.
B. In some kinds of matter, the outer electrons can barely move at all. These materials are
called insulators. Glass and rubber are typically
good insulators.
c. <u>Seniconductors</u> are materials that be made to sometimes behave as insulators and other times behave as conductors.
D. <u>Superconductors</u> are materials whose electrons may (at low temperatures, for example) roam entirely free, with no resistance whatsoever.

VI. Induction

When a charged object is brought near a neutral object, the neutral object will undergo "separation of charges". This is called charging by

## VII. Charging by Friction and Contact: Triboelectricity

For reasons that are not entirely understood, when two materials are brought into contact and then separated, one may become positively charged, while the other becomes negatively charged. Electricity created in this way is called Friko electricity.

The separated charges that result are often called <u>Static electricity</u>, because they do not move (until they are brought into contact with another object of different charge).

## Activity: "Sticky Tape" and Triboelectricity

- 1. Hang two pieces of tape from the edge of the table. Let them hang there and age. These will be called the "old tapes."
- 2. Attach a "base tape" to the table. Then stick another tape on top of it, folding over one end as a handle. Label it "B" (for bottom) or remember that it's the "b" tape.
- 3. Stick another tape on top of the "b" tape. Make a handle on the same end as the "b" tape handle. Label this one "T" (top).
- 4. Grab the B tape and pull up both tapes. Then pull them apart. Hold them back to back so they don't stick. What happens? Do they attract or repel? Fill in each open cell in the table on the right with "A" for "attract" or "R" for "repel."